# SAFETY DATA SHEET TersOx™ Buffer - Carbonate Mixture



Revision date: 2022-04-08 Version 1.1

# 1. PRODUCT AND COMPANY IDENTIFICATION

### **Product Identifier**

Product Name: TersOx™ Buffer - Carbonate Mixture

Synonyms: Sodium bicarbonate, Sodium carbonate

Product Form: Mixture

# Recommended use of the chemical and restrictions on use

Recommended Use: For use in buffering acid buildup in soil, sludge, and groundwater bioremediation

Restrictions on Use: Use as recommended by the label

# Details of the supplier and of the safety data sheet

Supplier Tersus Environmental, LLC

1116 Colonial Club Rd Wake Forest, NC 27587 Phone: +1-919-453-5577 Email: info@tersusenv.com

Contact Person David F. Alden

Phone: +1-919-453-5577 x2002 Email: <u>david.alden@tersusenv.com</u>

### **Emergency telephone number**

For leak, fire, spill or accident emergencies, call:

- +1-919-453-5577 (Tersus Office Hours, 8:00 AM to 5:00 PM Eastern)
- +1-800-424-9300 (Chemtrec 24 Hour Service Emergency Only)
- +1-703-527-3887 (Chemtrec Outside United States 24 Hour Service Emergency Only)
- +1-919-638-7892 Gary M. Birk (Outside office hours)

# 2. HAZARD IDENTIFICATION

# **Classification**

OSHA Regulatory Status This material is considered hazardous by the OSHA Hazard Communication Standard (29 CFR 1910.1200).

Serious eye damage/eye irritation Category 2B Specific target organ toxicity (single exposure) Category 2 Oxidizing Solids Category 1

# GHS Label elements, including precautionary statements

#### Label elements

Hazard Pictograms







Signal word

Danger

**Hazard Statements** 

Causes skin irritation (H315) Causes eye irritation (H320) May be corrosive to metals (H290)

**Precautionary statements - Prevention** 

P234 – Keep only in original container P264 – Wash thoroughly after handling P280 – Wear protective gloves

**Precautionary statements - Response** 

P302 + P352 - IF ON SKIN: Wash with soap and plenty of water.

P305 + P351 + P338 – IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P332 + P313 – If skin irritation occurs: Get medical advice/attention.
P337 - P313 – If eye irritation persists: Get medical advice/attention.
P362 + P364 – Take off contaminated clothing and wash it before reuse.

P390 – Absorb spillage to prevent material damage.

**Precautionary statements - Storage** 

P406 - Store in corrosive resistant container

**Precautionary Statements - Disposal** 

P501 - Dispose of contents/ container to an approved waste disposal plant

Other hazards

N/A

Unknown acute toxicity

0% of the mixture consists of ingredient(s) of unknown toxicity

# 3. COMPOSITION/INFORMATION ON INGREDIENTS

Chemical Formula Mixture Na<sub>2</sub>CO<sub>3</sub> and NaHCO<sub>3</sub>

**Hazardous components** 

Chemical Name	CAS Number	Concentration (wt. %)
Sodium Carbonate	497-19-8	N/A
Sodium Bicarbonate	144-55-8	N/A
Other inorganic calcium		
compounds		

Synonyms are provided in Section 1.

Occupational exposure limits, if available, are listed in Section 8.

#### 4. **FIRST AID MEASURES**

General Information In all cases of doubt, or when symptoms persist, seek medical attention.

Never give anything by mouth to an unconscious person.

**Eye Contact** Mild to severe irritant to the eyes. May cause redness, irritation and/or

conjunctivitis. In case of contact with eyes, flush eyes with low pressure water for at least 15 minutes holding eyelids open. If irritation persists, seek

medical attention.

Skin Contact Mild to severe irritant of the skin. May cause intense destruction of abraded

> skin. It is recommended that prolonged direct contact with skin be avoided. In case of contact with skin, wash skin with water for 15 minutes. Remove

contaminated clothing and wash.

Inhalation May cause irritation, sore throat and coughing if inhaled. Avoid

> inappropriate handling which may result in dust generation. If inhaled, remove from contaminated area to fresh air. Report situation. Seek medical

attention if allergic response exhibited.

Ingestion Ingestion of material may cause corrosion of gastric mucosa with sore

> throat and pain. May cause distention of the stomach, possible rupture. Renal injury will occur over 1g/kg. If swallowed, drink 2-4 glasses of water. Induce vomiting or perform gastric lavage if large amounts are ingested.

Get medical attention.

Most important symptoms and effects,

both acute and delayed

Corrosion of gastric mucosa if ingested.

Indication of any immediate medical attention and special treatment needed

Treat symptomatically

#### FIRE-FIGHTING MEASURES 5.

Suitable Extinguishing

Media

Unsuitable extinguishing

media

**Explosion Data** 

Sensitivity to Mechanical

Impact

Sensitivity to Static

Discharge

Specific Hazards Arising

from the chemical or mixture

Special Fire Fighting

**Procedures** 

Not sensitive.

Dry chemical. Foam.

Not sensitive.

Contain water runoff. Negligible hazard when exposed to flame.

Dry chemical, carbon dioxide, chemical foam or water fog

Self-contained breathing apparatus recommended for fire fighters if

large amount is present.

#### **ACCIDENTAL RELEASE MEASURES** 6.

**Personal Precautions Environmental Precautions Methods for Containment** 

and Clean Up

Provide sufficient ventilation

Spilled product should be removed immediately to avoid formation of dust. Contain spill, sweep up avoiding airborne dust. Provide sufficient

ventilation. Avoid wash down except for small traces.

# 7. HANDLING AND STORAGE

Precautions for safe handling

• Avoid formation of dust. Provide adequate ventilation of the room when handling this product.

Hygiene measures
Conditions for safe storage.

Provide eyewash capability.

including any incompatibilities

None

# 8. EXPOSRE CONTROL / PERSONAL PROTECTION

#### **Control parameters**

Exposure guidelines, ingredients with workplace control parameters.

**Exposure** 

CAS No.	Ingredient	Source	Value
497-19-8	Sodium carbonate	OSHA	No Established Limit
(Na <sub>2</sub> C	(Na <sub>2</sub> CO <sub>3</sub> )	ACGIH	No Established Limit
144-55-8	Sodium bicarbonate	OSHA	PEL 10 mg/m³ (total)
	(NaHCO₃)	ACGIH	TLV: 10 mg/m <sup>3</sup>

# **Exposure Control**

Appropriate engineering

controls

Provide adequate ventilation. Where reasonably practicable this should be achieved using local exhaust ventilation and good general extraction. If these are not enough to maintain concentrations of particulates and any vapor below occupational exposure limits suitable respiratory protection must be worn.

Eye/face protection

Safety glasses with side shield or face shield, or chemical goggles.

Respiratory protection

Skin

Hygiene measures

Dust mask or respirator for particle removal (NIOSH)

Wear long-sleeve shirt, trousers, safety shoes, gloves (rubber or vinyl) Maintain good housekeeping. Avoid dusty conditions. Wash hands and exposed skin after contact. Avoid contact with food or food preparation surfaces. If exposure of food surfaces occurs, wash with germicidal detergent or chlorine bleach. Remove and wash contaminated clothing.

### 9. PHYSICAL AND CHEMICAL PROPERTIES

# Information on basic physical and chemical properties

**Appearance** White crystalline powder

Odor None
Odor threshold Not determined

pH Does Not Apply
Melting point / freezing point 260 °C (550 °F)
Initial boiling point and boiling range
Flash Point Does Not Apply
260 °C (550 °F)
Not Measured
Not Measured

Evaporation rate (Ether = 1)

Flammability (solid, gas) Not Applicable

Upper/lower flammability or explosive

<u>limits</u>

Lower Explosive Limit:
Upper Explosive Limit:
Vapor pressure (Pa)
Vapor Density
Specific Gravity
Specific Gravity
Solubility in Water
Partition coefficient n-octanol/water
(Log Kow)
Auto-ignition temperature

Not Measured
Not Measured

Auto-ignition temperatureNot MeasuredDecomposition temperatureNot MeasuredViscosity (cSt)Not Measured

# 10. STABILITY AND REACTIVITY

Reactivity
Possibility of hazardous
reactions
Conditions to avoid

- Stable under normal conditions
- None known
- Exposure to strong acid can cause violent reaction with evolution of heat and carbon dioxide.
- Exposure to hot aluminum can cause explosive reaction with evolution
- Exposure to ammonium and silver nitrate can cause explosive reaction upon heating
- Exposure to aromatic amine and a chloronitro compound can cause a reaction with evolution of heat.
- Exposure to 2,4 dinitrotoluene can increase explosiveness.
- Exposure to flourine can cause violent ignition.
- Exposure to burning lithium releases radioactive sodium.
- Exposure to phosphorus pentoxide can cause a highly exothermic reaction.
- Exposure to hot sodium sulfide can cause an explosive reaction on exposure to water.
- Exposure to sulfuric acid can cause a violent eruption.
- Exposure to 2,4,6 trinitrotoluene can reduce explosion temperature.
- Exposure to zinc can be corrosive.

Incompatible materials

Hazardous decomposition products

None known

None known

# 11. TOXICOLOGICAL INFORMATION

#### **Acute Toxicity**

Classification	Category	Hazard Description
Acute toxicity (oral)		Not Applicable
Acute toxicity (dermal)		Not Applicable
Acute toxicity (inhalation)		Not Applicable
Skin corrosion/irritation		Mild
Serious eye damage/irritation		Mild
Respiratory sensitization		Large amounts will require
		medical attention.
Skin sensitization		Not Applicable
Germ cell mutagenicity		Not Applicable

Classification	Category	Hazard Description
Carcinogenicity		Not Applicable
Reproductive toxicity		Not Applicable
STOT-single exposure		Not Applicable
STOT-repeated exposure		Not Applicable
Aspiration hazard		Not Applicable

# 12. ECOLOGICAL INFORMATION

# **Aquatic Ecotoxicity**

Product degrades in time releasing plant nutrients. Increases pH of media.

### Persistence and degradability

Readily biodegradable

# **Bioaccumulative potential**

Information not available.

#### **Mobility in Soil**

Information not available.

# Other adverse effects

Information not available.

# 13. DISPOSAL CONSIDERATIONS

**Waste Disposal Methods** 

Dispose of in accordance with current Federal, State, and Local regulations.

# 14. TRANSPORTATION INFORMATION

# **DOT**

Proper Shipping Name Chemicals not otherwise indexed (NOI) non-hazardous.

#### **UN Number**

Mixture not classified as Hazardous according to Regulation (EC) 1272/2008.

# 15. REGULATORY INFORMATION

#### **Regulatory Overview**

All ingredients used are listed on the USEPA TSCA Inventory list.

OSHA: Not hazardous under 29 CFR 1910.1200

# 16. OTHER INFORMATION

Mixture classified as not dangerous according to Regulation (EC) 1272/2008.

Observe employment restrictions for people.

Product is not listed with IARC, NTP, ACGIH or OSHA as a carcinogen.

Components not precisely identified are proprietary or non-hazardous. All chemical ingredients appear on the EPA TSCA inventory.

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End of Safety Data Sheet